

I₂/KI-Mediated Oxidative N–N Bond Formation for the Synthesis of 1,5-Fused 1,2,4-Triazoles from *N*-Aryl Amidines

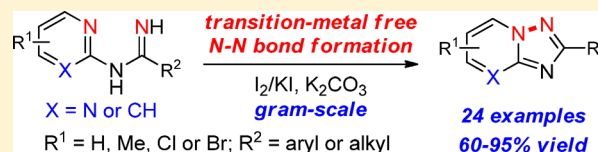
Lina Song,[†] Xianhai Tian,[†] Zhigang Lv,[†] Ertong Li,[†] Jie Wu,[†] Yangxue Liu,[†] Wenquan Yu,^{*,†} and Junbiao Chang^{*,†,‡}

[†]College of Chemistry and Molecular Engineering, Zhengzhou University, Zhengzhou, Henan 450001, P. R. China

[‡]Collaborative Innovation Center of New Drug Research and Safety Evaluation, Henan Province, Zhengzhou 450001, P. R. China

S Supporting Information

ABSTRACT: An I₂/KI-mediated oxidative N–N bond formation reaction is described. This new and environmentally benign approach allows for the convenient synthesis of a variety of 1,2,4-triazolo[1,5-*a*]pyridines and other 1,5-fused 1,2,4-triazoles from readily available *N*-aryl amidines in an efficient and scalable fashion.



1. INTRODUCTION

Owing to the numerous advantages associated with this eco-friendly element, molecular iodine plays an important role in organic synthesis.¹ As a catalyst, iodine has been extensively used in transformations, such as esterification, deprotection, Michael addition, and aldol reaction. It can also mediate iodocyclization, domino, and one-pot multicomponent reactions. More importantly, a number of oxidation reactions can be achieved by using iodine, for example, conversion of alcohols/aldehydes to esters, nitriles, or amides,^{1b} selective oxidation of alcohols to aldehydes and ketones,² and oxidative aromatization.^{1c} In particular, this environmentally benign oxidizing reagent has been successfully employed to construct C–C, C–N, C–O, and also C–S bonds.^{1h} However, to the best of our knowledge, there is no report of I₂-mediated N–N bond formation reactions. Encouraged by our previous work,³ in this paper, we envisioned the construction of N–N bonds by employing molecular iodine to synthesize the biologically important 1,2,4-triazole-fused heterocycles.⁴

Oxidative cyclization of *N*-(2-pyridyl) amidines is one of the most straightforward strategies for the construction of the 1,2,4-triazolo[1,5-*a*]pyridine framework, which previously has been achieved by utilizing oxidants, such as NaClO/base,⁵ Pb(OAc)₄,⁶ and MnO₂.⁷ Nevertheless, these methods are associated with some disadvantages, including low yields and limited scopes. In 2009, Ueda and Nagasawa⁸ reported a copper-catalyzed tandem addition–oxidative cyclization of 2-amino pyridines and aryl nitriles to 2-aryl-1,2,4-triazolo[1,5-*a*]pyridines. Alternatively, Zhao and co-workers⁹ developed a recyclable Cu–Zn/Al–Ti catalyst for the same transformation. In 2014, Du, Zhao, and co-workers¹⁰ reported a PIFA-mediated cyclization of *N*-(pyridin-2-yl) amidines to 2-aryl/2-alkyl triazolopyridines. Recently, Bartels, Fantasia, and co-workers¹¹ described a Cu-catalyzed aerobic oxidation of guanidylpyridines to the ones bearing 2-amino groups. Despite these elegant achievements made, it is still of importance to develop novel and general approaches to access this compound class. Herein, we disclose a new and efficient I₂/KI-mediated methodology

for the synthesis of both 2-aryl and 2-alkyl substituted 1,2,4-triazolo[1,5-*a*]pyridines, as well as their pyrazido- and pyrimidotriazole derivatives, from *N*-aryl amidines.

2. RESULTS AND DISCUSSION

The required substrates *N*-aryl amidines **2** were readily prepared via the addition reaction of corresponding aryl amines to substituted nitriles.^{8,10,11} Initial screening of a series of laboratory commonly used solvents suggested that dimethyl sulfoxide (DMSO) is the most suitable one for the I₂-mediated oxidative cyclization of substrates **2** to the fused [1,5-*a*]1,2,4-triazoles **1** in the presence of base (e.g., K₂CO₃). As shown in Table 1, full consumption of benzimidamide **2a** required 2.2 equiv of iodine, affording the expected product **1a** and its 6-iodinated derivative **1a'** (confirmed by X-ray; see the Supporting Information) in 52% and 47% yields, respectively (entry 1). When a methyl group was incorporated to the 5-position of the pyridine ring in the substrate, 1.5 equiv of iodine was enough for the complete conversion of **2b** to product **1b**, but still only in moderate yield with some unidentified byproducts formed (entry 2). The reaction also worked with weaker base (e.g., NaHCO₃), which, however, slowed down the reaction rate and decreased the yield (entry 3). During the optimization of the reaction conditions, we were pleased to observe that adding a catalytic amount of potassium iodide (KI) to the reaction system is favorable for the formation of product **1b** (entry 4). When a stoichiometric amount of KI was used, the yield was significantly increased up to 92% (entry 5). On this basis, lowering the temperature slowed down the reaction rate with no improvement in the yield (entry 6). Then, we started to investigate if KI would favor the formation of **1a** from substrate **2a**. In the presence of 1 equiv of KI, complete cyclization of **2a** needed only 1.2 equiv of iodine. The desired product **1a** was produced in an improved yield (79%), with less

Received: May 27, 2015

Published: June 26, 2015

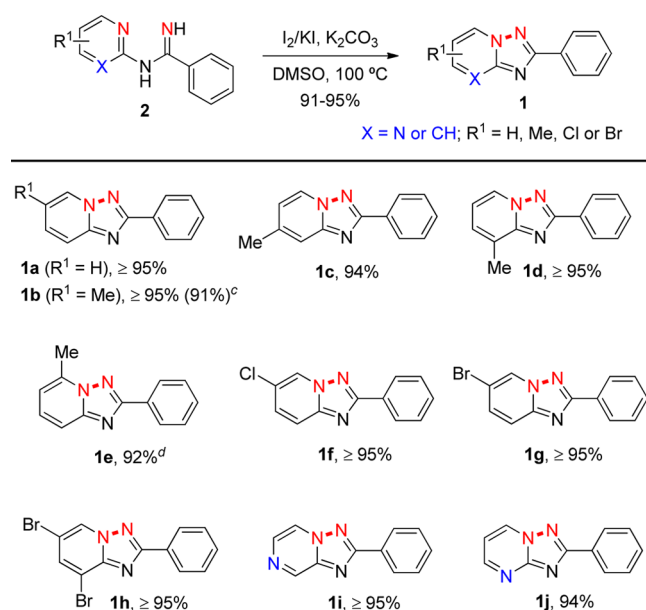
Table 1. Reaction Conditions Optimization for the Oxidative Cyclization of *N*-Aryl Amidines **2**^a

entry	substrate	I ₂ /equiv	KI/equiv	temp/°C	time/h	product, yield ^b
1	2a	2.2		100	1	1a, 52% 1a', 47%
2	2b	1.5		100	1	1b, 57%
3 ^c	2b	1.5		100	12	1b, 43%
4	2b	1.5	0.2	100	1	1b, 69%
5	2b	1.5	1	100	0.5	1b, 92%
6	2b	1.5	1	80	3.5	1b, 90%
7	2a	1.2	1	100	1	1a, 79% 1a', 20%
8	2a	1.2	5	100	2	1a, ≥95% 1a', trace
9	2a	1.2	1.5	100	1	1a, ≥95%
10	2b	1.2	1.5	100	1	1b, ≥95%

^aOptimal reaction conditions (entries 9–10): A well-stirred mixture of I₂ (0.6 mmol) and KI (0.75 mmol) in DMSO (5 mL) was treated with substrate **2** (0.5 mmol), followed by the addition of K₂CO₃ (1.5 mmol), and then heated to 100 °C for 1 h. ^bIsolated yields. ^cThe same amount of NaHCO₃ was used instead of K₂CO₃.

byproduct **1a'** formed (20%, entry 7). Increasing the dosage of potassium iodide to 5 equiv resulted in the formation of **1a** in excellent yield with only a trace amount of **1a'** observed (entry 8). Nevertheless, excessive KI has a negative impact on the conversion rate (entry 8), as it is also a byproduct formed during this oxidative cyclization process. In view of the above results, we assume that I₂ and KI may form a complex (e.g., KI₃),^{2,12} which can mediate this oxidative cyclization process more efficiently. On the basis of this hypothesis, we further optimized the experimental protocol by adding the substrates (**2a** or **2b**) to a well-stirred mixture of I₂ (1.2 equiv) and KI (1.5 equiv) in DMSO. To our delight, both the reactions finished within 1 h and afforded the desired product in excellent yields (entries 9–10). It is noteworthy that this oxidative cyclization reaction is insensitive to air and moisture. Taking substrate **2b** as an example, it was safely conducted on a gram scale (Scheme 1).

Then, a range of *N*-aryl benzimidamides **2** were subjected to the above optimal cyclization conditions to probe the reaction scope and generality (Scheme 1). All the substrates with methyl groups at the different positions of the pyridyl moiety (**2b–e**) were efficiently converted to the corresponding 2-phenyl-1,2,4-triazolo[1,5-*a*]pyridines (**1b–e**). The steric block effect of the 6-methyl group in **2e** could be responsible for the slightly decreased yield of the product (**1e**), and the completeness of this transformation also required a relatively longer reaction time. Substrates bearing electron-withdrawing groups on the pyridine rings (**2f–h**) were also cyclized to the desired products **1f–h**. Additionally, pyrazido- (**1i**) and pyrimidotriazoles (**1j**) were prepared via the oxidative cyclization of *N*-pyrazyl (**2i**) and *N*-pyrimidyl substituted benzimidamides (**2j**), respectively, in good yields.

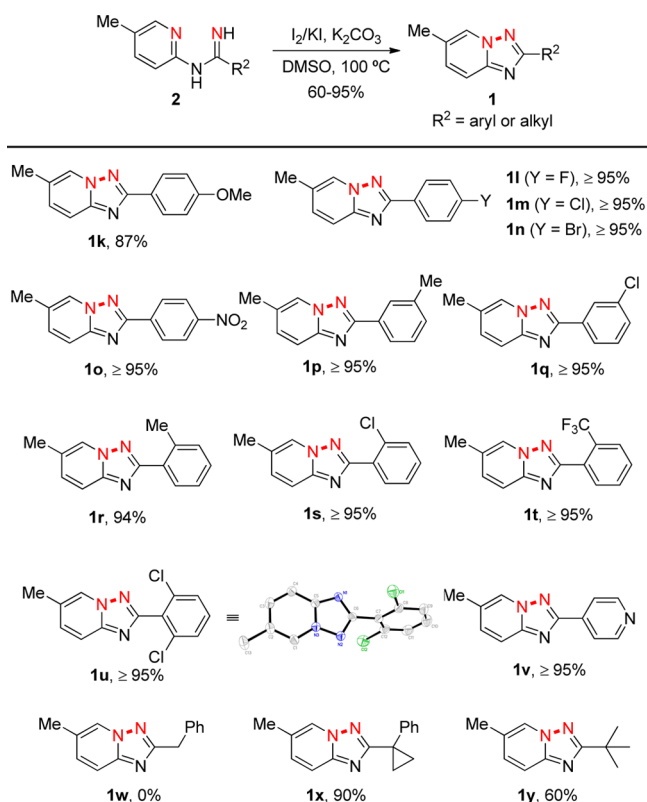
Scheme 1. Substrate Scope for the Synthesis of 2-Phenyl Fused [1,5-*a*]1,2,4-Triazoles **1**^{a,b}

^aOptimal reaction conditions: A well-stirred mixture of I₂ (0.6 mmol) and KI (0.75 mmol) in DMSO (5 mL) was treated with substrate **2** (0.5 mmol), followed by the addition of K₂CO₃ (1.5 mmol), and then heated to 100 °C for 1 h. ^bIsolated yields. ^cYield of gram-scale reaction (6 mmol) in parentheses. ^dThe reaction time was 1.5 h.

In light of these encouraging results, we initiated further studies by replacing the phenyl moiety in substrate **2b** with a variety of aromatic and aliphatic substituents (R², Scheme 2). When R² is an aryl group, this methodology is compatible with both electron-donating and electron-withdrawing groups at *para*-, *meta*-, and *ortho*-positions of the benzene ring (**1k–t**). Even the nitro-group-bearing substrate **2o** was smoothly transformed into the desired 1,2,4-triazolo[1,5-*a*]pyridine **1o** under the optimal reaction conditions. *Ortho*-substitution on the 2-phenyl moiety did not affect either the reaction rate or the yields of the products (**1r–u**). The 2,6-dichlorophenyl substrate (**2u**) was completely cyclized to the expected product **1t** within 1 h in excellent yield, of which the structure is further confirmed by X-ray crystallography (see the Supporting Information). Moreover, 2-pyridyl triazolopyridine (**1v**) was obtained in decent yield from isonicotinimidamide **2v**. This protocol is also amenable to the amidines formed from aliphatic nitriles (R²CN); however, the presence of α -hydrogens of the R² group may prevent the generation of the desired product, as illustrated in the case of **2w** to **1w**. On the other hand, compound **2x** and **2y** were successfully converted into the target products **1x–y**. Noticeably, the sensitive cyclopropane ring in substrate **2x** was untouched under the present cyclization conditions.

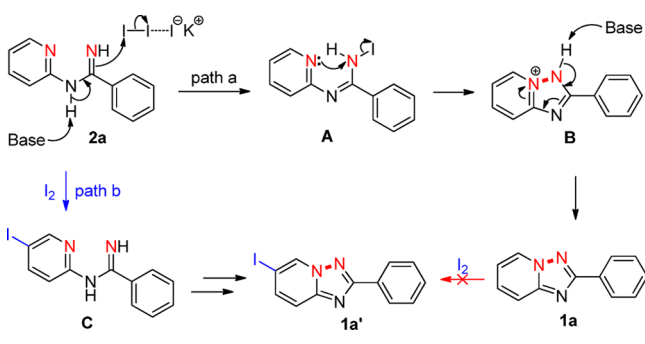
A plausible mechanism for the oxidative cyclization process of *N*-aryl amidines **2** is proposed (Scheme 3). Taking the formation of **1a** as an example, the base-promoted oxidative iodination of substrate **2a** gives an iodide intermediate **A** (path a). Then, the N–I bond in iodide **A** cleaves, and consequently, an ammonium ion **B** is generated via an S_N2'-type cyclization of **A** with a new N–N bond formed. Finally, the subsequent deprotonation and rearomatization afford the 1,2,4-triazolo[1,5-*a*]pyridine framework **1a**. To investigate the formation of the

Scheme 2. Substrate Scope for the Synthesis of 2-Substituted 1,2,4-Triazolo[1,5-*a*]pyridines **1^{a,b}**



^aOptimal reaction conditions: A well-stirred mixture of I₂ (0.6 mmol) and KI (0.75 mmol) in DMSO (5 mL) was treated with substrate **2** (0.5 mmol), followed by the addition of K₂CO₃ (1.5 mmol), and then heated to 100 °C for 1 h. ^bIsolated yields.

Scheme 3. Proposed Mechanism for the Formation of **1a and **1a'****



iodinated byproduct **1a'**, a control experiment was conducted by further treating **1a** with iodine in the presence of K₂CO₃ in DMSO at 100 °C. It turned out that no reaction occurred at all under the above reaction conditions. This result demonstrated that the formation of **1a'** did not undergo **1a** and the iodination on the pyridine ring should take place before the cyclization process (path b). When KI was added to the reaction system, it will complex with I₂ to form KI₃,^{2,12} which can mediate the oxidative cyclization of substrate **2a** to the desired product **1a** (path a) more efficiently than free molecular iodine by diminishing the iodination (path b) and other side reactions.

3. CONCLUSIONS

In summary, for the first time, an I₂-mediated oxidative N–N bond formation reaction has been developed for the synthesis of 1,5-fused 1,2,4-triazoles. Addition of KI to the reaction system successfully suppressed the iodination and other side reactions. This facile and transition-metal-free synthetic process works well with a wide range of *N*-aryl substituted amidines and can be safely conducted on a gram scale. The features such as generality, high efficiency, short reaction time, and air- and moisture-insensitive reaction conditions make the present method an attractive alternative for the preparation of 1,2,4-triazolo[1,5-*a*]pyridines and other fused 1,2,4-triazole derivatives.

4. EXPERIMENTAL SECTION

4.1. General Information. ¹H and ¹³C NMR spectra were recorded on a 400 MHz (100 MHz for ¹³C NMR) spectrometer. Chemical shift values are given in ppm (parts per million) with tetramethylsilane (TMS) as an internal standard. The peak patterns are indicated as follows: s, singlet; d, doublet; t, triplet; q, quartet; quint, quintet; sext, sextet; m, multiplet; dd, doublet of doublets; dt, doublet of triplets. The coupling constants (*J*) are reported in hertz (Hz). Melting points were determined on a micromelting point apparatus without corrections. High-resolution mass spectra (HRMS-ESI) were obtained on a Q-TOF mass spectrometer. For the preparation of substrates **2**, analytical grade reagents *N,N*-dimethylformamide (DMF) and dimethyl sulfoxide (DMSO) were dried over activated MS 4 Å prior to use; for the oxidative cyclization of **2** to **1**, analytical grade reagent DMSO was used without further treatment. Flash column chromatography was performed over silica gel 200–300 mesh, and the eluent was a mixture of ethyl acetate (EA) and petroleum ether (PE).

4.2. Preparation of Substrates **2.** **4.2.1. General Procedure A.⁸** A solution of substituted 2-aminopyridine (10 mmol) in DMF (5 mL) was treated with NaH (60% dispersion in mineral oil, 0.6 g, 15 mmol) at 0 °C and stirred at the same temperature for 30 min. The corresponding nitrile (R²CN, 15 mmol) was then added to the reaction mixture, which was stirred at room temperature until TLC indicated the total consumption of the substituted 2-aminopyridine. The reaction was quenched with 5% aqueous NaHCO₃ (20 mL) and extracted with EA (3 × 30 mL). The combined organic layers was washed with brine (40 mL), dried over Na₂SO₄, concentrated, and then purified through silica gel column chromatography using a mixture of EA and PE as the eluent enriched with 1% of triethylamine to afford substrates **2a–g**, **2m–n**, **2q**, and **2s**.

4.2.2. General Procedure B.¹⁰ The nitrile (R²CN, 15 mmol) was taken in a dry sealed tube, to which was added substituted 2-aminopyridine (10 mmol) under a stream of nitrogen. The contents in the sealed tube were stirred at 90 °C for 30 min. The above mixture was treated with SnCl₄ (1.76 mL, 15 mmol) and heated to 110 °C for 5 h. After cooling to room temperature, the resulting solid was crushed into powder and dissolved in hot water. The aqueous suspension was made alkaline (pH > 11) with 2 M aqueous NaOH and extracted with CH₂Cl₂ (3 × 100 mL). The combined organic layer was dried over anhydrous Na₂SO₄, concentrated, and then purified through silica gel column chromatography using a mixture of EA and PE as the eluent enriched with 1% of triethylamine to afford substrates **2h**, **2k–l**, **2o–p**, **2r**, and **2t–y**.

4.2.3. General Procedure C.¹¹ Under a stream of nitrogen, potassium *tert*-butoxide (2.47 g, 22 mmol) was added to a well-stirred solution of 2-pyrazineamine (or 2-pyrimidineamine, 0.95 g, 10 mmol) in anhydrous DMSO (5 mL). After the mixture was stirred at room temperature for 15 min, benzonitrile (2.06 g, 20 mmol) was added dropwise to the above mixture. The reaction mixture was stirred at room temperature for another 30 min and then heated to 50 °C until TLC indicated the total consumption of the aryl amine. After cooling to room temperature, the reaction was quenched with cold water (20 mL), and extracted with CH₂Cl₂ (3 × 30 mL). The combined organic

layer was dried over anhydrous Na_2SO_4 , concentrated, and then purified through silica gel column chromatography using a mixture of EA and PE as the eluent enriched with 1% of triethylamine to afford substrates 2i–j.

N-(Pyridin-2-yl)benzimidamide (2a). Following General Procedure A, 2a was obtained as a white solid: yield 1.44 g, 73%; mp 96–97 °C (lit.⁸ mp 96–97 °C); $R_f = 0.24$ (EA/PE 20:80); $^1\text{H NMR}$ (400 MHz, CD_3OD) δ 8.34 (dd, $J = 5.2, 1.6$ Hz, 1H), 7.87–7.85 (m, 2H), 7.73–7.69 (m, 1H), 7.52–7.45 (m, 3H), 7.16 (d, $J = 8.0$ Hz, 1H), 7.01–6.98 (m, 1H); $^{13}\text{C NMR}$ (400 MHz, CD_3OD) δ 162.3, 161.4, 146.0, 137.6, 136.9, 130.4, 128.1, 127.2, 120.2, 117.8; HRMS (m/z) [$M + H$]⁺ calcd for $\text{C}_{12}\text{H}_{12}\text{N}_3$ 198.1026, found 198.1030.

N-(5-Methylpyridin-2-yl)benzimidamide (2b). Following General Procedure A, 2b was obtained as a white solid: yield 1.88 g, 89%; mp 86–87 °C (lit.¹³ mp 89–90 °C); $R_f = 0.24$ (EA/PE 20:80); $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.16–8.15 (m, 1H), 7.91–7.89 (m, 2H), 7.48–7.43 (m, 4H), 7.20 (d, $J = 8.0$ Hz, 1H), 2.29 (s, 3H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 160.7, 158.4, 145.8, 138.4, 137.6, 130.4, 128.5, 127.0, 126.9, 121.9, 18.0; HRMS (m/z) [$M + H$]⁺ calcd for $\text{C}_{13}\text{H}_{14}\text{N}_3$ 212.1182, found 212.1188.

N-(4-Methylpyridin-2-yl)benzimidamide (2c). Following General Procedure A, 2c was obtained as a white solid: yield 1.54 g, 73%; mp 124–125 °C (lit.¹³ mp 125–126 °C); $R_f = 0.23$ (EA/PE 20:80); $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.19 (d, $J = 5.2$ Hz, 1H), 7.91 (dd, $J = 7.2, 1.6$ Hz, 2H), 7.45–7.44 (m, 3H), 7.12 (s, 1H), 6.77 (dd, $J = 5.2, 1.2$ Hz, 1H), 2.33 (s, 3H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 162.9, 159.0, 148.5, 145.6, 137.5, 130.4, 128.5, 126.9, 122.8, 119.2, 21.0; HRMS (m/z) [$M + H$]⁺ calcd for $\text{C}_{13}\text{H}_{14}\text{N}_3$ 212.1182, found 212.1185.

N-(3-Methylpyridin-2-yl)benzimidamide (2d). Following General Procedure A, 2d was obtained as a white solid: yield 1.54 g, 73%; mp 68–69 °C (lit.¹³ mp 68–69 °C); $R_f = 0.44$ (EA/PE 20:80); $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.18 (dd, $J = 4.8, 1.6$ Hz, 1H), 8.00–7.97 (m, 2H), 7.50–7.44 (m, 4H), 6.84 (dd, $J = 7.2, 4.8$ Hz, 1H), 2.48 (s, 3H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 161.1, 157.2, 143.5, 138.0, 137.5, 130.5, 130.4, 128.5, 127.0, 117.8, 18.5; HRMS (m/z) [$M + H$]⁺ calcd for $\text{C}_{13}\text{H}_{14}\text{N}_3$ 212.1182, found 212.1187.

N-(6-Methylpyridin-2-yl)benzimidamide (2e). Following General Procedure A, 2e was obtained as a light yellow solid: yield 1.50 g, 71%; mp 63–65 °C; $R_f = 0.26$ (EA/PE 20:80); $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 7.91–7.89 (m, 2H), 7.55 (t, $J = 8.0$ Hz, 1H), 7.46–7.44 (m, 3H), 7.09 (d, $J = 8.0$ Hz, 1H), 6.80 (d, $J = 7.2$ Hz, 1H), 2.51 (s, 3H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 162.4, 158.9, 154.7, 137.8, 137.6, 130.4, 128.6, 126.9, 119.3, 117.2, 24.5; HRMS (m/z) [$M + H$]⁺ calcd for $\text{C}_{13}\text{H}_{14}\text{N}_3$ 212.1182, found 212.1187.

N-(5-Chloropyridin-2-yl)benzimidamide (2f). Following General Procedure A, 2f was obtained as a white solid: yield 1.11 g, 48%; mp 155–156 °C (lit.¹⁰ mp 157–158 °C); $R_f = 0.39$ (EA/PE 20:80); $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.27 (d, $J = 2.4$ Hz, 1H), 7.91–7.89 (m, 2H), 7.59 (dd, $J = 8.8, 2.8$ Hz, 1H), 7.49–7.43 (m, 3H), 7.22 (d, $J = 8.8$ Hz, 1H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 161.2, 159.1, 144.4, 137.3, 137.1, 130.8, 128.6, 126.9, 125.1, 123.5; HRMS (m/z) [$M + H$]⁺ calcd for $\text{C}_{12}\text{H}_{11}\text{ClN}_3$ 232.0636, found 232.0647.

N-(5-Bromopyridin-2-yl)benzimidamide (2g). Following General Procedure A, 2g was obtained as a white solid: yield 1.44 g, 52%; mp 158–159 °C; $R_f = 0.39$ (EA/PE 20:80); $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.37 (d, $J = 2.8$ Hz, 1H), 7.91–7.89 (m, 2H), 7.72 (dd, $J = 8.4, 2.4$ Hz, 1H), 7.48–7.43 (m, 3H), 7.16 (d, $J = 8.8$ Hz, 1H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 161.5, 159.2, 146.7, 140.0, 137.1, 130.8, 128.6, 126.9, 124.1, 113.2; HRMS (m/z) [$M + H$]⁺ calcd for $\text{C}_{12}\text{H}_{11}\text{BrN}_3$ 276.1031, found 276.1034.

N-(3,5-Dibromopyridin-2-yl)benzimidamide (2h). Following General Procedure B, 2h was obtained as a white solid: yield 2.92 g, 82%; mp 125–126 °C (lit.¹⁰ mp 128–129 °C); $R_f = 0.32$ (EA/PE 89:11); $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.30 (m, 1H), 8.06–8.02 (m, 3H), 7.53–7.45 (m, 3H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 159.1, 157.9, 145.6, 142.9, 136.4, 131.3, 128.7, 127.3, 119.1, 111.9; HRMS (m/z) [$M + H$]⁺ calcd for $\text{C}_{12}\text{H}_{10}\text{Br}_2\text{N}_3$ 355.9216, found 355.9228.

N-(Pyrazin-2-yl)benzimidamide (2i). Following General Procedure C, 2i was obtained as a white solid: yield 1.40 g, 71%; mp 160–161 °C; $R_f = 0.25$ (EA/PE 25:75); $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 10.28

(br, s, 1H), 8.62 (d, $J = 1.2$ Hz, 1H), 8.22 (dd, $J = 2.8, 1.6$ Hz, 1H), 8.14 (d, $J = 2.8$ Hz, 1H), 7.96–7.93 (m, 2H), 7.51–7.45 (m, 3H), 6.22 (br, s, 1H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 160.9, 158.7, 145.8, 139.7, 137.1, 136.6, 131.1, 128.7, 127.0; HRMS (m/z) [$M + H$]⁺ calcd for $\text{C}_{11}\text{H}_{11}\text{N}_4$ 199.0978, found 199.0979.

N-(Pyrimidin-2-yl)benzimidamide (2j). Following General Procedure C, 2j was obtained as a white solid: yield 1.33 g, 67%; mp 148–149 °C (lit.¹⁴ mp 149 °C); $R_f = 0.27$ (EA/PE 99:1); $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.63 (d, $J = 4.8$ Hz, 2H), 8.04–8.02 (m, 2H), 7.49–7.41 (m, 3H), 6.89 (t, $J = 4.8$ Hz, 1H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 157.4, 136.3, 131.1, 128.5, 127.3, 114.6; HRMS (m/z) [$M + H$]⁺ calcd for $\text{C}_{11}\text{H}_{11}\text{N}_4$ 199.0978, found 199.0989.

4-Methoxy-N-(5-methylpyridin-2-yl)benzimidamide (2k). Following General Procedure B, 2k was obtained as a white solid: yield 2.08 g, 86%; mp 118–119 °C; $R_f = 0.32$ (EA/PE 99:1); $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.14 (d, $J = 2.4$ Hz, 1H), 7.88–7.86 (m, 2H), 7.46 (dd, $J = 8.4, 2.4$ Hz, 1H), 7.17 (d, $J = 8.4$ Hz, 1H), 6.95–6.93 (m, 2H), 3.84 (s, 3H), 2.29 (s, 3H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 156.7, 156.1, 153.2, 141.0, 133.6, 125.1, 123.7, 122.0, 117.0, 109.0, 50.6, 13.2; HRMS (m/z) [$M + H$]⁺ calcd for $\text{C}_{14}\text{H}_{16}\text{N}_3\text{O}$ 242.1287, found 242.1296.

4-Fluoro-N-(5-methylpyridin-2-yl)benzimidamide (2l). Following General Procedure B, 2l was obtained as a white solid: yield 1.67 g, 73%; mp 149–151 °C; $R_f = 0.25$ (EA/PE 20:80); $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.15 (d, $J = 2.4$ Hz, 1H), 7.92–7.83 (m, 2H), 7.47 (dd, $J = 8.4, 2.4$ Hz, 1H), 7.17 (d, $J = 8.4$ Hz, 1H), 7.13–7.09 (m, 2H), 2.30 (s, 3H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 164.2 (d, $J_{\text{C-F}} = 248.8$ Hz), 160.6, 157.3, 145.8, 138.5, 133.7, 129.0 (d, $J_{\text{C-F}} = 8.6$ Hz), 127.2, 121.9, 115.4 (d, $J_{\text{C-F}} = 21.6$ Hz), 18.0; HRMS (m/z) [$M + H$]⁺ calcd for $\text{C}_{13}\text{H}_{13}\text{FN}_3$ 230.1088, found 230.1093.

4-Chloro-N-(5-methylpyridin-2-yl)benzimidamide (2m). Following General Procedure A, 2m was obtained as a white solid: yield 1.87 g, 76%; mp 167–168 °C; $R_f = 0.28$ (EA/PE 20:80); $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.16–8.15 (m, 1H), 7.86–7.83 (m, 2H), 7.49–7.46 (m, 1H), 7.42–7.38 (m, 2H), 7.17 (d, $J = 8.4$ Hz, 1H), 2.30 (s, 3H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 160.5, 157.1, 145.8, 138.5, 136.4, 128.7, 128.3, 127.3, 122.0, 18.0; HRMS (m/z) [$M + H$]⁺ calcd for $\text{C}_{13}\text{H}_{13}\text{ClN}_3$ 246.0793, found 246.0795.

4-Bromo-N-(5-methylpyridin-2-yl)benzimidamide (2n). Following General Procedure A, 2n was obtained as a white solid: yield 2.21 g, 76%; mp 172–174 °C; $R_f = 0.20$ (EA/PE 20:80); $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.15 (d, $J = 2.0$ Hz, 1H), 7.79–7.77 (m, 2H), 7.58–7.55 (m, 2H), 7.48 (dd, $J = 8.0, 2.4$ Hz, 1H), 7.17 (d, $J = 8.4$ Hz, 1H), 2.30 (s, 3H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 160.5, 157.2, 145.8, 138.5, 136.4, 131.7, 128.5, 127.3, 124.8, 122.0, 18.0; HRMS (m/z) [$M + H$]⁺ calcd for $\text{C}_{13}\text{H}_{13}\text{BrN}_3$ 290.0287, found 290.0295.

N-(5-Methylpyridin-2-yl)-4-nitrobenzimidamide (2o). Following General Procedure B, 2o was obtained as a yellow solid: yield 2.00 g, 78%; mp 147–148 °C; $R_f = 0.31$ (EA/PE 25:75); $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.29–8.26 (m, 2H), 8.19–8.18 (m, 1H), 8.09–8.07 (m, 2H), 7.53–7.50 (m, 1H), 7.20 (d, $J = 8.4$ Hz, 1H), 2.32 (s, 3H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 160.1, 155.9, 148.9, 145.9, 143.4, 138.7, 128.1, 128.0, 123.7, 122.3, 18.1; HRMS (m/z) [$M + H$]⁺ calcd for $\text{C}_{13}\text{H}_{13}\text{N}_4\text{O}_2$ 257.1033, found 257.1039.

3-Methyl-N-(5-methylpyridin-2-yl)benzimidamide (2p). Following General Procedure B, 2p was obtained as a white solid: yield 2.12 g, 94%; mp 54–55 °C; $R_f = 0.28$ (EA/PE 25:75); $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.15 (d, $J = 2.4$ Hz, 1H), 7.75 (s, 1H), 7.65 (d, $J = 7.6$ Hz, 1H), 7.47 (dd, $J = 8.4, 2.4$ Hz, 1H), 7.32 (t, $J = 7.6$ Hz, 1H), 7.27 (d, $J = 7.6$ Hz, 1H), 7.19 (d, $J = 8.4$ Hz, 1H), 2.41 (s, 3H), 2.30 (s, 3H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 160.8, 158.6, 145.7, 138.4, 138.3, 137.5, 131.2, 128.4, 127.7, 127.0, 123.7, 121.9, 21.5, 18.0; HRMS (m/z) [$M + H$]⁺ calcd for $\text{C}_{14}\text{H}_{16}\text{N}_3$ 226.1339, found 226.1339.

3-Chloro-N-(5-methylpyridin-2-yl)benzimidamide (2q). Following General Procedure A, 2q was obtained as a white solid: yield 2.21 g, 90%; mp 129–130 °C; $R_f = 0.29$ (EA/PE 20:80); $^1\text{H NMR}$ (400 MHz, CDCl_3) δ 8.16 (d, $J = 2.4$ Hz, 1H), 7.93 (t, $J = 1.6$ Hz, 1H), 7.75 (dt, $J = 7.6, 1.2$ Hz, 1H), 7.48 (dd, $J = 8.4, 2.4$ Hz, 1H), 7.44–7.41 (m, 1H), 7.36 (t, $J = 8.0$ Hz, 1H), 7.19 (d, $J = 8.4$ Hz, 1H), 2.30 (s, 3H); $^{13}\text{C NMR}$ (100 MHz, CDCl_3) δ 160.4, 156.9, 145.8, 139.4, 138.5,

134.6, 130.4, 129.8, 127.4, 127.3, 124.8, 122.1, 18.0; HRMS (m/z) [$M + H$]⁺ calcd for C₁₃H₁₃ClN₃ 246.0793, found 246.0806.

2-Methyl-*N*-(5-methylpyridin-2-yl)benzimidamide (2r). Following General Procedure B, **2r** was obtained as a white solid: yield 0.93 g, 41%; mp 121–122 °C; R_f = 0.27 (EA/PE 99:1); ¹H NMR (400 MHz, CDCl₃) δ 10.47 (br, s, 1H), 8.16 (d, J = 1.6 Hz, 1H), 7.48–7.44 (m, 2H), 7.31–7.21 (m, 3H), 7.12 (d, J = 8.4 Hz, 1H), 5.51 (br, s, 1H), 2.51 (s, 3H), 2.30 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 160.9, 160.6, 145.8, 138.7, 138.6, 135.6, 130.9, 129.2, 127.8, 127.3, 126.0, 121.8, 19.8, 18.1; HRMS (m/z) [$M + H$]⁺ calcd for C₁₄H₁₆N₃ 226.1339, found 226.1348.

2-Chloro-*N*-(5-methylpyridin-2-yl)benzimidamide (2s). Following General Procedure A, **2s** was obtained as a white solid: yield 1.57 g, 64%; mp 132–133 °C; R_f = 0.20 (EA/PE 20:80); ¹H NMR (400 MHz, CDCl₃) δ 8.16 (d, J = 1.6 Hz, 1H), 7.68 (dd, J = 5.6, 3.6 Hz, 1H), 7.47 (dd, J = 8.4, 2.4 Hz, 1H), 7.42–7.40 (m, 1H), 7.35–7.30 (m, 2H), 7.15 (d, J = 8.0 Hz, 1H), 2.30 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 160.6, 157.8, 145.7, 138.5, 137.3, 131.3, 130.44, 130.36, 130.1, 127.4, 127.1, 121.8, 18.0; HRMS (m/z) [$M + H$]⁺ calcd for C₁₃H₁₃ClN₃ 246.0793, found 246.0793.

***N*-(5-Methylpyridin-2-yl)-2-(trifluoromethyl)benzimidamide (2t).** Following General Procedure B, **2t** was obtained as a white solid: yield 1.37 g, 49%; mp 162–163 °C; R_f = 0.28 (EA/PE 25:75); ¹H NMR (400 MHz, CDCl₃) δ 10.55 (br, s, 1H), 8.17 (d, J = 1.6 Hz, 1H), 7.72 (d, J = 7.6 Hz, 1H), 7.69 (d, J = 7.6 Hz, 1H), 7.61 (t, J = 7.6 Hz, 1H), 7.52 (t, J = 7.6 Hz, 1H), 7.47 (dd, J = 8.0, 2.4 Hz, 1H), 7.13 (d, J = 8.4 Hz, 1H), 5.59 (br, s, 1H), 2.31 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 160.5, 158.2, 145.8, 138.5, 137.3, 132.1, 130.0, 129.2, 127.7 (q, J_{C-F} = 31.3 Hz), 127.6, 126.4 (q, J_{C-F} = 5.0 Hz), 123.9 (q, J_{C-F} = 272.4 Hz), 121.8, 18.0; HRMS (m/z) [$M + H$]⁺ calcd for C₁₄H₁₃F₃N₃ 280.1056, found 280.1064.

2,6-Dichloro-*N*-(5-methylpyridin-2-yl)benzimidamide (2u). Following General Procedure B, **2u** was obtained as a white solid: yield 2.13 g, 76%; mp 179–181 °C; R_f = 0.20 (EA/PE 20:80); ¹H NMR (400 MHz, CDCl₃) δ 8.18 (d, J = 2.0 Hz, 1H), 7.48 (dd, J = 8.4, 2.4 Hz, 1H), 7.36–7.34 (m, 2H), 7.26–7.22 (m, 1H), 7.16 (d, J = 8.4 Hz, 1H), 2.31 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 160.4, 155.1, 145.9, 138.5, 136.5, 133.6, 130.2, 128.2, 127.8, 121.8, 18.0; HRMS (m/z) [$M + H$]⁺ calcd for C₁₃H₁₂Cl₂N₃ 280.0403, found 280.0403.

***N*-(5-Methylpyridin-2-yl)isonicotinimidamide (2v).** Following General Procedure B, **2v** was obtained as a white solid: yield 1.82 g, 86%; mp 144–145 °C; R_f = 0.30 (EA/PE 99:1); ¹H NMR (400 MHz, DMSO-*d*₆) δ 8.72 (d, J = 6.0 Hz, 2H), 8.22 (s, 1H), 7.97 (d, J = 5.6 Hz, 2H), 7.58 (dd, J = 8.4, 2.0 Hz, 1H), 7.10 (d, J = 8.0 Hz, 1H), 2.27 (s, 3H); ¹³C NMR (100 MHz, DMSO-*d*₆) δ 160.7, 155.6, 150.4, 146.4, 144.2, 139.0, 127.4, 122.0, 121.7, 17.9; HRMS (m/z) [$M + H$]⁺ calcd for C₁₂H₁₃N₄ 213.1135, found 213.1134.

***N*-(5-Methylpyridin-2-yl)-2-phenylacetimidamide (2w).** Following General Procedure B, **2w** was obtained as a white solid: yield 1.92 g, 85%; mp 109–110 °C; R_f = 0.28 (EA/CH₃OH 83:17); ¹H NMR (400 MHz, CDCl₃) δ 8.07 (d, J = 2.4 Hz, 1H), 7.44 (dd, J = 8.4, 2.4 Hz, 1H), 7.36–7.34 (m, 4H), 7.32–7.27 (m, 1H), 7.07 (d, J = 8.0 Hz, 1H), 3.75 (s, 2H), 2.26 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 161.4, 160.6, 145.7, 138.5, 136.2, 129.5, 128.9, 127.2, 126.8, 120.9, 44.9, 17.9; HRMS (m/z) [$M + H$]⁺ calcd for C₁₄H₁₆N₃ 226.1339, found 226.1353.

***N*-(5-Methylpyridin-2-yl)-1-phenylcyclopropanecarboximidamide (2x).** Following General Procedure B, **2x** was obtained as a colorless oil: yield 2.18 g, 87%; R_f = 0.23 (EA/PE 34:66); ¹H NMR (400 MHz, CDCl₃) δ 8.02 (m, 1H), 7.47–7.45 (m, 2H), 7.39–7.33 (m, 3H), 7.30–7.25 (m, 1H), 6.97 (d, J = 8.0 Hz, 1H), 2.23 (s, 3H), 1.77–1.75 (m, 2H), 1.15–1.12 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 164.3, 160.9, 145.7, 140.9, 138.3, 130.7, 128.8, 127.5, 126.1, 120.9, 30.8, 17.9, 16.0; HRMS (m/z) [$M + H$]⁺ calcd for C₁₆H₁₈N₃ 252.1495, found 252.1496.

***N*-(5-Methylpyridin-2-yl)pivalimidamide (2y).** Following General Procedure B, **2y** was obtained as a colorless oil: yield 0.27 g, 14%; R_f = 0.30 (EA/PE 50:50); ¹H NMR (400 MHz, CDCl₃) δ 8.08 (m, 1H), 7.42 (dd, J = 8.4, 2.4 Hz, 1H), 7.15 (d, J = 8.0 Hz, 1H), 2.26 (s, 3H), 1.33 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 169.1, 145.7, 138.5,

126.8, 120.6, 113.4, 37.9, 28.5, 17.9; HRMS (m/z) [$M + H$]⁺ calcd for C₁₁H₁₈N₃ 192.1495, found 192.1502.

4.3. Synthesis of Products 1. **4.3.1. Synthesis of 2-Phenyl-[1,2,4]triazolo[1,5-*a*]pyridine (1a) and 7-Iodo-2-phenyl-[1,2,4]triazolo[1,5-*a*]pyridine (1a').** A stirred solution of *N*-(pyridin-2-yl)benzimidamide (**2a**, 99 mg, 0.5 mmol) in DMSO (5.0 mL) was treated with iodine (279 mg, 1.1 mmol) and K₂CO₃ (311 mg, 2.25 mmol). The reaction mixture was heated to 100 °C for 1 h (TLC indicated that the conversion was complete). After cooling to room temperature, it was quenched with 5% Na₂S₂O₃ (0.5 mL), followed by the addition of brine (15 mL), and then extracted with EA (3 × 15 mL). The combined organic layer was dried over anhydrous Na₂SO₄, concentrated, and purified through column chromatography using a mixture of EA and PE as the eluent to afford products **1a** (51 mg, 52%, white solid) and **1a'** (76 mg, 47%, white solid).

7-Iodo-2-phenyl-[1,2,4]triazolo[1,5-*a*]pyridine (1a'). White solid, mp 176–178 °C; R_f = 0.25 (EA/PE 10:90); ¹H NMR (400 MHz, CDCl₃) δ 8.86 (m, 1H), 8.27–8.25 (m, 2H), 7.69 (dd, J = 9.2, 1.6 Hz, 1H), 7.55 (dd, J = 9.6, 0.8 Hz, 1H), 7.51–7.48 (m, 3H); ¹³C NMR (400 MHz, CDCl₃) δ 164.3, 150.6, 137.7, 133.4, 130.3, 130.2, 128.7, 127.3, 117.2, 75.5; HRMS (m/z) [$M + H$]⁺ calcd for C₁₂H₉N₃I 321.9836, found 321.9832.

4.3.2. Synthesis of 1,2,4-Triazolo[1,5-*a*]pyridines 1 (General Procedure D). A mixture of KI (125 mg, 0.75 mmol) and iodine (152 mg, 0.6 mmol) in DMSO (5 mL) was stirred at room temperature for 10 min and then treated with *N*-substituted amidine **2** (0.5 mmol), followed by the addition of K₂CO₃ (208 mg, 1.5 mmol). The reaction mixture was heated to 100 °C until TLC indicated the total consumption of the substrate **2** (1–1.5 h). After cooling to room temperature, the reaction was quenched with 5% Na₂S₂O₃ (0.5 mL), followed by the addition of brine (15 mL), and then extracted with EA (3 × 15 mL). The combined organic layers were dried over anhydrous Na₂SO₄, concentrated, and then purified through silica gel column chromatography using a mixture of EA and PE as the eluent to afford the desired product **1**.

2-Phenyl-[1,2,4]triazolo[1,5-*a*]pyridine (1a). Following General Procedure D, **1a** was obtained as a white solid: yield 97 mg, ≥95%; mp 134–136 °C (lit.⁸ mp 134–135 °C); R_f = 0.25 (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.59 (dt, J = 6.8, 1.2 Hz, 1H), 8.31–8.28 (m, 2H), 7.75 (dt, J = 8.8, 0.8 Hz, 1H), 7.52–7.46 (m, 4H), 6.99 (dt, J = 1.2, 6.8 Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 164.1, 151.7, 130.7, 130.1, 129.5, 128.7, 128.3, 127.3, 116.4, 113.6; HRMS (m/z) [$M + H$]⁺ calcd for C₁₂H₁₀N₃ 196.0869, found 196.0873.

6-Methyl-2-phenyl-[1,2,4]triazolo[1,5-*a*]pyridine (1b). Following General Procedure D, **1b** was obtained as a white solid: 0.5 mmol scale, yield 103 mg, ≥95%; 6 mmol scale, yield 1.138 g, 91%. mp 118–119 °C (lit.¹⁵ mp 120.5–121.0 °C); R_f = 0.28 (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.38 (s, 1H), 8.28–8.26 (m, 2H), 7.64 (d, J = 9.2 Hz, 1H), 7.51–7.45 (m, 3H), 7.33 (dd, J = 9.2, 1.6 Hz, 1H), 2.40 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 163.8, 150.3, 132.3, 130.9, 129.9, 128.6, 127.1, 126.3, 123.7, 115.5, 18.1; HRMS (m/z) [$M + H$]⁺ calcd for C₁₃H₁₂N₃ 210.1026, found 210.1034.

7-Methyl-2-phenyl-[1,2,4]triazolo[1,5-*a*]pyridine (1c). Following General Procedure D, **1c** was obtained as a white solid: yield 98 mg, 94%; mp 141–142 °C (lit.¹⁶ mp 140–142 °C); R_f = 0.28 (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.44 (d, J = 7.2 Hz, 1H), 8.28–8.26 (m, 2H), 7.51–7.45 (m, 4H), 6.80 (dd, J = 7.2, 2.0 Hz, 1H), 2.47 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 164.2, 151.9, 141.0, 130.9, 129.9, 128.6, 127.2, 116.0, 115.0, 21.6; HRMS (m/z) [$M + H$]⁺ calcd for C₁₃H₁₂N₃ 210.1026, found 210.1029.

8-Methyl-2-phenyl-[1,2,4]triazolo[1,5-*a*]pyridine (1d). Following General Procedure D, **1d** was obtained as a white solid: yield 100 mg, ≥95%; mp 96–97 °C (lit.¹⁰ mp 100–101 °C); R_f = 0.44 (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.43 (d, J = 6.8 Hz, 1H), 8.31–8.29 (m, 2H), 7.51–7.45 (m, 3H), 7.26–7.22 (m, 1H), 6.87 (t, J = 6.8 Hz, 1H), 2.68 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 163.7, 152.2, 131.0, 129.9, 128.6, 128.1, 127.3, 127.0, 125.9, 113.4, 17.0; HRMS (m/z) [$M + H$]⁺ calcd for C₁₃H₁₂N₃ 210.1026, found 210.1033.

5-Methyl-2-phenyl-[1,2,4]triazolo[1,5-*a*]pyridine (1e). Following General Procedure D, **1e** was obtained as a white solid: yield 96 mg, 92%; mp 84–85 °C (lit.¹⁷ mp 83–85 °C); $R_f = 0.33$ (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.34–8.32 (m, 2H), 7.63 (d, $J = 8.4$ Hz, 1H), 7.52–7.40 (m, 4H), 6.81 (dt, $J = 7.2, 0.8$ Hz, 1H), 2.84 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 163.5, 151.8, 138.9, 131.2, 129.9, 129.3, 128.6, 127.3, 113.5, 112.8, 17.6; HRMS (m/z) [M + H]⁺ calcd for C₁₃H₁₂N₃ 210.1026, found 210.1025.

6-Chloro-2-phenyl-[1,2,4]triazolo[1,5-*a*]pyridine (1f). Following General Procedure D, **1f** was obtained as a white solid: yield 114 mg, ≥95%; mp 165–167 °C (lit.¹⁰ mp 168–169 °C); $R_f = 0.47$ (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.64 (s, 1H), 8.27–8.25 (m, 2H), 7.69 (d, $J = 9.6$ Hz, 1H), 7.52–7.46 (m, 4H); ¹³C NMR (100 MHz, CDCl₃) δ 164.9, 150.3, 130.9, 130.34, 130.31, 128.7, 127.3, 126.6, 121.5, 116.4; HRMS (m/z) [M + H]⁺ calcd for C₁₂H₉ClN₃ 230.0479, found 230.0481.

6-Bromo-2-phenyl-[1,2,4]triazolo[1,5-*a*]pyridine (1g). Following General Procedure D, **1g** was obtained as a white solid: yield 136 mg, ≥95%; mp 165–167 °C (lit.¹⁰ mp 165–166 °C); $R_f = 0.32$ (EA/PE 11:89); ¹H NMR (400 MHz, CDCl₃) δ 8.74 (d, $J = 1.6$ Hz, 1H), 8.27–8.25 (m, 2H), 7.66–7.63 (m, 1H), 7.58–7.55 (m, 1H), 7.52–7.47 (m, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 164.7, 150.4, 133.1, 130.34, 130.29, 128.8, 128.7, 127.3, 116.8, 107.7; HRMS (m/z) [M + H]⁺ calcd for C₁₂H₉BrN₃ 275.9955, found 275.9955.

6,8-Dibromo-2-phenyl-[1,2,4]triazolo[1,5-*a*]pyridine (1h).¹⁰ Following General Procedure D, **1h** was obtained as a white solid: yield 176 mg, ≥95%; mp 149–151 °C; $R_f = 0.32$ (EA/PE 5:95); ¹H NMR (400 MHz, CDCl₃) δ 8.70 (d, $J = 1.6$ Hz, 1H), 8.30–8.28 (m, 2H), 7.84 (d, $J = 1.6$ Hz, 1H), 7.50–7.48 (m, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 164.8, 149.7, 135.0, 130.6, 129.8, 128.7, 127.9, 127.6, 110.1, 106.9; HRMS (m/z) [M + H]⁺ calcd for C₁₂H₈Br₂N₃ 353.9059, found 353.9056.

2-Phenyl-[1,2,4]triazolo[1,5-*a*]pyridine (1i). Following General Procedure D, **1i** was obtained as a white solid: yield 98 mg, ≥95%; mp 179–180 °C; $R_f = 0.33$ (EA/PE 25:75); ¹H NMR (400 MHz, CDCl₃) δ 9.30 (d, $J = 1.2$ Hz, 1H), 8.56 (dd, $J = 4.8, 1.6$ Hz, 1H), 8.32–8.30 (m, 2H), 8.18 (d, $J = 4.4$ Hz, 1H), 7.54–7.51 (m, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 165.1, 147.3, 142.7, 131.6, 130.8, 129.8, 128.9, 127.6, 121.5; HRMS (m/z) [M + H]⁺ calcd for C₁₁H₉N₄ 197.0822, found 197.0822.

2-Phenyl-[1,2,4]triazolo[1,5-*a*]pyrimidine (1j). Following General Procedure D, **1j** was obtained as a white solid: yield 92 mg, 94%; mp 184–185 °C (lit.¹⁴ mp 184 °C); $R_f = 0.33$ (EA/PE 50:50); ¹H NMR (400 MHz, CDCl₃) δ 8.86 (dd, $J = 6.4, 2.0$ Hz, 1H), 8.80 (dd, $J = 4.4, 2.0$ Hz, 1H), 8.36–8.34 (m, 2H), 7.53–7.50 (m, 3H), 7.09 (dd, $J = 6.8, 4.4$ Hz, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 166.3, 156.1, 154.4, 135.5, 130.8, 130.2, 128.8, 127.5, 110.0; HRMS (m/z) [M + Na]⁺ calcd for C₁₁H₈N₄Na 219.0641, found 219.0656.

2-(4-Methoxyphenyl)-6-methyl-[1,2,4]triazolo[1,5-*a*]pyridine (1k). Following General Procedure D, **1k** was obtained as a white solid: yield 104 mg, 87%; mp 166–167 °C; $R_f = 0.26$ (EA/PE 25:75); ¹H NMR (400 MHz, CDCl₃) δ 8.36–8.35 (m, 1H), 8.22–8.18 (m, 2H), 7.61 (d, $J = 9.2$ Hz, 1H), 7.32 (dd, $J = 8.8, 1.6$ Hz, 1H), 7.03–6.99 (m, 2H), 3.87 (s, 3H), 2.40 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 163.7, 161.0, 150.3, 132.3, 128.6, 126.2, 123.5, 123.4, 115.3, 114.0, 55.3, 18.1; HRMS (m/z) [M + H]⁺ calcd for C₁₄H₁₄N₃O 240.1131, found 240.1142.

2-(4-Fluorophenyl)-6-methyl-[1,2,4]triazolo[1,5-*a*]pyridine (1l). Following General Procedure D, **1l** was obtained as a white solid: yield 113 mg, ≥95%; mp 163–164 °C; $R_f = 0.22$ (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.36 (m, 1H), 8.27–8.22 (m, 2H), 7.62 (d, $J = 8.8$ Hz, 1H), 7.34 (dd, $J = 9.2, 1.6$ Hz, 1H), 7.19–7.14 (m, 2H), 2.40 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 163.9 (d, $J_{C-F} = 247.8$ Hz), 163.0, 150.3, 132.4, 129.1 (d, $J_{C-F} = 8.4$ Hz), 127.2 (d, $J_{C-F} = 3.1$ Hz), 126.3, 123.8, 115.7 (d, $J_{C-F} = 21.6$ Hz), 115.5, 18.0; HRMS (m/z) [M + H]⁺ calcd for C₁₃H₁₁FN₃ 228.0932, found 228.0938.

2-(4-Chlorophenyl)-6-methyl-[1,2,4]triazolo[1,5-*a*]pyridine (1m). Following General Procedure D, **1m** was obtained as a white solid: yield 119 mg, ≥95%; mp 169–171 °C; $R_f = 0.28$ (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.36 (s, 1H), 8.21–8.18 (m, 2H), 7.63 (d,

$J = 9.2$ Hz, 1H), 7.46–7.44 (m, 2H), 7.35 (dd, $J = 9.2, 1.6$ Hz, 1H), 2.41 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 162.8, 150.3, 135.9, 132.5, 129.5, 128.9, 128.4, 126.3, 123.9, 115.5, 18.1; HRMS (m/z) [M + H]⁺ calcd for C₁₃H₁₁ClN₃ 244.0636, found 244.0644.

2-(4-Bromophenyl)-6-methyl-[1,2,4]triazolo[1,5-*a*]pyridine (1n). Following General Procedure D, **1n** was obtained as a white solid: yield 143 mg, ≥95%; mp 179–181 °C; $R_f = 0.25$ (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.36 (s, 1H), 8.14–8.11 (m, 2H), 7.64–7.59 (m, 3H), 7.35 (dd, $J = 9.2, 1.6$ Hz, 1H), 2.41 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 162.9, 150.3, 132.5, 131.8, 130.0, 128.7, 126.3, 124.2, 123.9, 115.6, 18.1; HRMS (m/z) [M + H]⁺ calcd for C₁₃H₁₁BrN₃ 288.0131, found 288.0128.

6-Methyl-2-(4-nitrophenyl)-[1,2,4]triazolo[1,5-*a*]pyridine (1o). Following General Procedure D, **1o** was obtained as a light yellow solid: yield 126 mg, ≥95%; mp decomposed at 277–280 °C; $R_f = 0.25$ (EA/PE 17:83); ¹H NMR (400 MHz, DMSO-*d*₆) δ 8.89 (s, 1H), 8.45–8.37 (m, 4H), 7.84 (d, $J = 8.8$ Hz, 1H), 7.62 (dd, $J = 9.2, 1.6$ Hz, 1H), 2.41 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 153.3, 150.3, 141.8, 141.4, 132.5, 128.8, 128.7, 128.4, 124.5, 110.8, 16.2; HRMS (m/z) [M + H]⁺ calcd for C₁₃H₁₁N₄O₂ 255.0876, found 255.0875.

6-Methyl-2-(*m*-tolyl)-[1,2,4]triazolo[1,5-*a*]pyridine (1p). Following General Procedure D, **1p** was obtained as a white solid: yield 111 mg, ≥95%; mp 129–131 °C; $R_f = 0.27$ (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.38–8.37 (m, 1H), 8.10 (s, 1H), 8.07 (d, $J = 7.6$ Hz, 1H), 7.64 (d, $J = 9.2$ Hz, 1H), 7.38 (t, $J = 7.6$ Hz, 1H), 7.33 (dd, $J = 9.2, 1.6$ Hz, 1H), 7.27 (d, $J = 7.2$ Hz, 1H), 2.45 (s, 3H), 2.40 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 163.9, 150.3, 138.4, 132.4, 130.8, 130.7, 128.6, 127.8, 126.3, 124.3, 123.7, 115.5, 21.4, 18.1; HRMS (m/z) [M + H]⁺ calcd for C₁₄H₁₄N₃ 224.1182, found 224.1182.

2-(3-Chlorophenyl)-6-methyl-[1,2,4]triazolo[1,5-*a*]pyridine (1q). Following General Procedure D, **1q** was obtained as a white solid: yield 121 mg, ≥95%; mp 153 °C; $R_f = 0.31$ (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.37 (s, 1H), 8.28–8.27 (m, 1H), 8.17–8.12 (m, 1H), 7.64 (d, $J = 8.8$ Hz, 1H), 7.42–7.41 (m, 2H), 7.36 (dd, $J = 9.2, 1.6$ Hz, 1H), 2.41 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 162.5, 150.3, 134.7, 132.8, 132.6, 129.9, 129.8, 127.3, 126.3, 125.2, 124.1, 115.6, 18.1; HRMS (m/z) [M + H]⁺ calcd for C₁₃H₁₁ClN₃ 244.0636, found 244.0639.

6-Methyl-2-(*o*-tolyl)-[1,2,4]triazolo[1,5-*a*]pyridine (1r). Following General Procedure D, **1r** was obtained as a white solid: yield 105 mg, 94%; mp 95–96 °C; $R_f = 0.35$ (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.41 (m, 1H), 8.06–8.03 (m, 1H), 7.66 (d, $J = 9.2$ Hz, 1H), 7.36–7.30 (m, 4H), 2.70 (s, 3H), 2.42 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 160.0, 144.8, 132.9, 127.4, 126.4, 125.6, 125.4, 124.5, 121.5, 121.0, 118.7, 110.8, 17.0, 13.3; HRMS (m/z) [M + H]⁺ calcd for C₁₄H₁₄N₃ 224.1182, found 224.1194.

2-(2-Chlorophenyl)-6-methyl-[1,2,4]triazolo[1,5-*a*]pyridine (1s). Following General Procedure D, **1s** was obtained as a white solid: yield 121 mg, ≥95%; mp 109–110 °C; $R_f = 0.22$ (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.45 (s, 1H), 8.00–7.98 (m, 1H), 7.70 (d, $J = 9.2$ Hz, 1H), 7.55–7.52 (m, 1H), 7.39–7.37 (m, 3H), 2.43 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 162.5, 149.6, 133.1, 132.5, 132.0, 130.7, 130.5, 130.2, 126.7, 126.4, 124.0, 115.8, 18.1; HRMS (m/z) [M + H]⁺ calcd for C₁₃H₁₁ClN₃ 244.0636, found 244.0647.

6-Methyl-2-(2-(trifluoromethyl)phenyl)-[1,2,4]triazolo[1,5-*a*]pyridine (1t). Following General Procedure D, **1t** was obtained as a white solid: yield 138 mg, ≥95%; mp 80 °C; $R_f = 0.31$ (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.44–8.43 (m, 1H), 7.88 (d, $J = 7.6$ Hz, 1H), 7.84 (d, $J = 7.2$ Hz, 1H), 7.70 (d, $J = 9.2$ Hz, 1H), 7.65 (t, $J = 7.6$ Hz, 1H), 7.59 (t, $J = 7.6$ Hz, 1H), 7.40 (dd, $J = 9.2, 2.0$ Hz, 1H), 2.44 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 162.7, 149.7, 132.6, 132.3, 131.5, 130.5 (q, $J_{C-F} = 2.0$ Hz), 129.4, 129.2 (q, $J_{C-F} = 31.4$ Hz), 126.6 (q, $J_{C-F} = 5.4$ Hz), 126.4, 124.1, 123.8 (q, $J_{C-F} = 272.1$ Hz), 115.9, 18.1; HRMS (m/z) [M + H]⁺ calcd for C₁₄H₁₁F₃N₃ 278.0899, found 278.0909.

2-(2,6-Dichlorophenyl)-6-methyl-[1,2,4]triazolo[1,5-*a*]pyridine (1u). Following General Procedure D, **1u** was obtained as a white solid: yield 138 mg, ≥95%; mp 171 °C; $R_f = 0.28$ (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.45 (s, 1H), 7.73 (d, $J = 9.2$ Hz, 1H), 7.44–7.41 (m, 3H), 7.36–7.32 (m, 1H), 2.45 (s, 3H); ¹³C NMR (100

MHz, CDCl₃) δ 160.2, 149.8, 136.0, 132.6, 131.0, 130.9, 127.9, 126.6, 124.2, 116.1, 18.1; HRMS (*m/z*) [M + H]⁺ calcd for C₁₃H₁₀Cl₂N₃, 278.0246, found 278.0246.

6-Methyl-2-(pyridin-4-yl)-[1,2,4]triazolo[1,5-a]pyridine (1v). Following General Procedure D, **1v** was obtained as a white solid: yield 105 mg, $\geq 95\%$; mp 181–182 °C; *R*_f = 0.25 (EA/PE 67:33); ¹H NMR (400 MHz, CDCl₃) δ 8.76–8.75 (m, 2H), 8.40 (s, 1H), 8.13–8.11 (m, 2H), 7.68 (d, *J* = 8.8 Hz, 1H), 7.40 (dd, *J* = 8.8, 1.6 Hz, 1H), 2.44 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 161.5, 150.4, 138.4, 132.9, 126.4, 124.6, 121.2, 115.9, 18.1; HRMS (*m/z*) [M + H]⁺ calcd for C₁₂H₁₁N₄, 211.0978, found 211.0978.

6-Methyl-2-(1-phenylcyclopropyl)-[1,2,4]triazolo[1,5-a]pyridine (1x). Following General Procedure D, **1x** was obtained as a white solid: yield 112 mg, 90%; mp 77–78 °C; *R*_f = 0.25 (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.23–8.22 (m, 1H), 7.52–7.49 (m, 3H), 7.38–7.34 (m, 2H), 7.30–7.24 (m, 2H), 2.33 (s, 3H), 1.71–1.68 (m, 2H), 1.41–1.39 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 170.3, 149.8, 142.2, 132.0, 130.1, 128.4, 126.9, 126.2, 122.8, 115.0, 25.2, 18.0, 16.5; HRMS (*m/z*) [M + H]⁺ calcd for C₁₆H₁₆N₃, 250.1339, found 250.1341.

2-(tert-Butyl)-6-methyl-[1,2,4]triazolo[1,5-a]pyridine (1y). Following General Procedure D, **1y** was obtained as a white solid: yield 57 mg, 60%; mp 69–70 °C; *R*_f = 0.30 (EA/PE 17:83); ¹H NMR (400 MHz, CDCl₃) δ 8.32–8.31 (m, 1H), 7.58 (d, *J* = 9.2 Hz, 1H), 7.30 (dd, *J* = 8.8, 1.6 Hz, 1H), 2.38 (s, 3H), 1.48 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 174.4, 149.7, 131.9, 126.2, 122.9, 115.2, 33.2, 29.7, 18.0; HRMS (*m/z*) [M + H]⁺ calcd for C₁₁H₁₆N₃, 190.1339, found 190.1340.

■ ASSOCIATED CONTENT

● Supporting Information

Copies of ¹H and ¹³C NMR spectra of compounds **1** and **2**, and X-ray structures and data of compounds **1a'** and **1u** (CIF). The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.joc.5b01183.

■ AUTHOR INFORMATION

Corresponding Authors

*E-mail: changjunbiao@zzu.edu.cn (J.C.).

*E-mail: wenquan_yu@zzu.edu.cn (W.Y.).

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

We thank the National Natural Science Foundation of China (Nos. 81302637 and 81330075) and the China Postdoctoral Science Foundation (Nos. 2013M530341 and 2014T70690) for financial support.

■ REFERENCES

(1) For recent reviews on application of iodine in organic synthesis, see: (a) Banerjee, A. K.; Vera, W.; Mora, H.; Laya, M. S.; Bedoya, L.; Cabrera, E. V. *J. Sci. Ind. Res.* **2006**, *65*, 299. (b) Togo, H.; Iida, S. *Synlett* **2006**, 2006, 2159. (c) Finkbeiner, P.; Nachtsheim, B. J. *Synthesis* **2013**, *45*, 979. (d) Ren, Y.-M.; Cai, C.; Yang, R.-C. *RSC Adv.* **2013**, *3*, 7182. (e) Mphahlele, M. J. *Molecules* **2009**, *14*, 5308. (f) Jereb, M.; Vražič, D.; Zupan, M. *Tetrahedron* **2011**, *67*, 1355. (g) Parvatkar, P. T.; Parameswaran, P. S.; Tilve, S. G. *Chem. - Eur. J.* **2012**, *18*, 5460. (h) Zhao, J.; Gao, W.; Chang, H.; Li, X.; Liu, Q.; Wei, W. *Youji Huaxue* **2014**, *34*, 1941 (in Chinese).

(2) Gogoi, P.; Konwar, D. *Org. Biomol. Chem.* **2005**, *3*, 3473.

(3) For I₂-mediated oxidative C–X (X = O, S and N) bond formation reactions developed by our groups, see: (a) Yu, W.; Huang, G.; Zhang, Y.; Liu, H.; Dong, L.; Yu, X.; Li, Y.; Chang, J. *J. Org. Chem.* **2013**, *78*, 10337. (b) Niu, P.; Kang, J.; Tian, X.; Song, L.; Liu, H.; Wu, J.; Yu, W.; Chang, J. *J. Org. Chem.* **2015**, *80*, 1018. (c) Zhang, X.; Kang, J.; Niu, P.; Wu, J.; Yu, W.; Chang, J. *J. Org. Chem.* **2014**, *79*, 10170.

(4) (a) Menet, C. J.; Fletcher, S. R.; Van Lommen, G. V.; Geney, R.; Blanc, J.; Smits, K.; Jouannigot, N.; Deprez, P.; van der Aar, E. M.; Clement-Lacroix, P.; Lepescheux, L.; Galien, R.; Vayssiere, B.; Nelles, L.; Christophe, T.; Brys, R.; Uhring, M.; Ciesielski, F.; Van Rompaey, L. V. *J. Med. Chem.* **2014**, *57*, 9323. (b) Marwaha, A.; White, J.; El Mazouni, F.; Creason, S. A.; Kokkonda, S.; Buckner, F. S.; Charman, S. A.; Phillips, M. A.; Rathod, P. K. *J. Med. Chem.* **2012**, *55*, 7425. (c) Wang, X.-M.; Xu, J.; Li, Y.-P.; Li, H.; Jiang, C.-S.; Yang, G.-D.; Lu, S.-M.; Zhang, S.-Q. *Eur. J. Med. Chem.* **2013**, *67*, 243. (d) Oguro, Y.; Cary, D. R.; Miyamoto, N.; Tawada, M.; Iwata, H.; Miki, H.; Hori, A.; Imamura, S. *Bioorg. Med. Chem.* **2013**, *21*, 4714. (e) Bell, K.; Sunose, M.; Ellard, K.; Cansfield, A.; Taylor, J.; Miller, W.; Ramsden, N.; Bergamini, G.; Neubauer, G. *Bioorg. Med. Chem. Lett.* **2012**, *22*, 5257. (f) Sunose, M.; Bell, K.; Ellard, K.; Bergamini, G.; Neubauer, G.; Werner, T.; Ramsden, N. *Bioorg. Med. Chem. Lett.* **2012**, *22*, 4613. (g) Siu, M.; Pastor, R.; Liu, W.; Barrett, K.; Berry, M.; Blair, W. S.; Chang, C.; Chen, J. Z.; Eigenbrot, C.; Ghilardi, N.; Gibbons, P.; He, H.; Hurley, C. A.; Kenny, J. R.; Khojasteh, S. C.; Le, H.; Lee, L.; Lyssikatos, J. P.; Magnuson, S.; Pulk, R.; Tsui, V.; Ultsch, M.; Xiao, Y.; Zhu, B.-Y.; Sampath, D. *Bioorg. Med. Chem. Lett.* **2013**, *23*, 5014. (h) Bergamini, G.; Bell, K.; Shimamura, S.; Werner, T.; Cansfield, A.; Müller, K.; Perrin, J.; Rau, C.; Ellard, K.; Hopf, C.; Doce, C.; Leggate, D.; Mangano, R.; Mathieson, T.; O'Mahony, A.; Plavec, I.; Rharbaoui, F.; Reinhard, F.; Savitski, M. M.; Ramsden, N.; Hirsch, E.; Drewes, G.; Rausch, O.; Bantscheff, M.; Neubauer, G. *Nat. Chem. Biol.* **2012**, *8*, 576. (i) Dugan, B. J.; Gingrich, D. E.; Mesaros, E. F.; Milkiewicz, K. L.; Curry, M. A.; Zulli, A. L.; Dobrzanski, P.; Serdikoff, C.; Jan, M.; Angeles, T. S.; Albom, M. S.; Mason, J. L.; Aimone, L. D.; Meyer, S. L.; Huang, Z.; Wells-Knecht, K. J.; Ator, M. A.; Ruggeri, B. A.; Dorsey, B. D. *J. Med. Chem.* **2012**, *55*, 5243.

(5) Grenda, V. J.; Jones, R. E.; Gal, G.; Sletzing, M. *J. Org. Chem.* **1965**, *30*, 259.

(6) Potts, K. T.; Burton, H. R.; Bhattacharyya, J. *J. Org. Chem.* **1966**, *31*, 260.

(7) Raval, J. P.; Desai, K. R. *ARKIVOC* **2005**, 2005, 21.

(8) Ueda, S.; Nagasawa, H. *J. Am. Chem. Soc.* **2009**, *131*, 15080.

(9) Meng, X.; Yu, C.; Zhao, P. *RSC Adv.* **2014**, *4*, 8612.

(10) Zheng, Z.; Ma, S.; Tang, L.; Zhang-Negrerie, D.; Du, Y.; Zhao, K. *J. Org. Chem.* **2014**, *79*, 4687.

(11) Bartels, B.; Bolas, C. G.; Cueni, P.; Fantasia, S.; Gaeng, N.; Trita, A. S. *J. Org. Chem.* **2015**, *80*, 1249.

(12) Svensson, P. H.; Kloos, L. *Chem. Rev.* **2003**, *103*, 1649.

(13) Potts, K. T.; Burton, H. R.; Bhattacharyya, J. *J. Org. Chem.* **1966**, *31*, 260.

(14) Prasad, V. S. R.; Reddy, K. K. *Synth. Commun.* **1990**, *20*, 2617.

(15) Pauchard, J. P.; Siegrist, A. E. *Helv. Chim. Acta* **1978**, *61*, 142.

(16) Takehi, A.; Ito, S.; Uchiyama, K.; Konno, Y. *Chem. Lett.* **1976**, 413.

(17) Gilchrist, T. L.; Harris, C. J.; Hawkins, D. G.; Moody, C. J.; Rees, C. W. *J. Chem. Soc., Perkin Trans. 1* **1976**, 2166.